

[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one

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Ring closure of 6-amino-3-oxo-*as*-triazine-5-thione with  $\alpha$ -halo ketones provides the thiazino[2,3-*e*]-1,2,4-triazines which dehydrate *via* an unusual pathway to give 7-aryl-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-ones.

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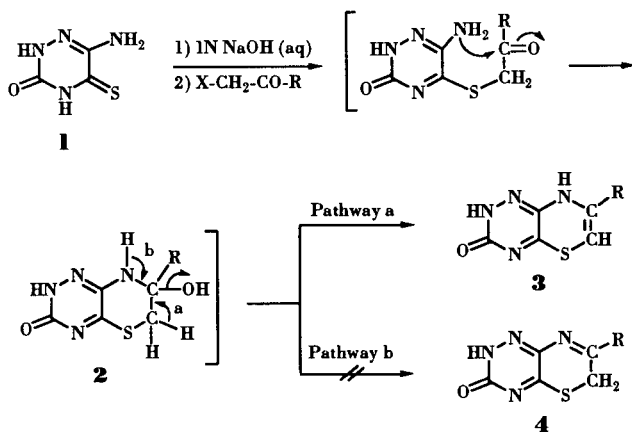
An examination of numerous antitumor agents from both natural and synthetic origin suggests that a common chemical structural pattern of the 2-phenylnaphthalene-type ring system is necessary for designing compounds of this type. The 2-phenylnaphthalene-type ring systems could either be carbocyclic or heterocyclic, with nitrogen, oxygen and/or sulfur atoms placed at selected positions [1]. We have synthesized a number of *as*-triazines [2-3] and its condensed heterocycles, pyrazino[2,3-*e*]-1,2,4-triazine [4] and imidazo[4,5-*e*]-1,2,4-triazine [5-6] and studied their chemical and biological properties. *as*-Triazine-3,5-dione (6-azauracil) has proved to possess a broad spectrum of therapeutic effects which include antiviral [7-8], antitumor [9-10], and antifungal activities [11]. The present report describes the preparation of another novel thiazino[2,3-*e*]-1,2,4-triazine which provides the first example of condensed *as*-triazine belonging to the antitumor 2-phenylnaphthalene-type ring system.

6-Amino-3-oxo-*as*-triazine-5-thione (**1**) [12], previously prepared in our laboratory, was reacted with  $\alpha$ -halo ke-

tones in the presence of sodium acetate as described by Kim *et al* [13]. This reaction resulted in the formation of complicated products which were difficult to separate. Faced with this dilemma, we explored a more effective route for the preparation of this heterocycle. It was found that when the sodium salt of 3-oxo-*as*-triazine-5-thione in dry ethanol was alkylated with an  $\alpha$ -halo ketone it cyclized *in situ* to provide thiazino[2,3-*e*]-1,2,4-triazine (**2**) as the sole product. Depending on the mode of dehydration of **2**, two tautomers are possible, *i.e.* 8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3**) and 6*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**4**).

According to Kim *et al* [13], reaction of 2,5,6-triamino-4-thiopyrimidine with  $\alpha$ -halo ketones gave substituted 7*H*-pyrimido[4,5-*b*][1,4]thiazines which resemble tautomer **4**. Nair [14] and Henrie [15] have also reported the preparation of 2-amino-4(3*H*)-oxo-6-phenylpyrimido[4,5-*b*][1,4]-thiazine from 2,5-diamino-6-mercaptopyrimidone and  $\alpha$ -bromoacetophenone. Faced with a similar situation, we could have reached the same conclusion, *i.e.*, that the dehydration product was **4** (pathway **b**, Scheme 1). However, after careful examination of the nmr spectra it became apparent that only **3** (pathway **a**) fit the data. The proton nmr spectra of the dehydration product showed a vinyl signal at 4.6-4.9 ppm and no evidence for allylic protons. The C-13 nmr showed only a tertiary and no methylene carbon lent additional support for the tautomer **3**. In order to provide a more definitive structural assignment, we decided to synthesize 7-methyl-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3e**) and examine its proton nmr. The long range coupling ( $J = 1.2$  Hz) of its vinylic proton and the protons of the methyl substituent further confirmed our assignment. Based on above nmr spectral data, we conclude that the dehydration occurs *via* pathway **a** to give tautomer **3**.

Scheme 1



- 3a**, R = C<sub>6</sub>H<sub>5</sub>
- 3b**, R = *p*-OCH<sub>3</sub>-C<sub>6</sub>H<sub>4</sub>
- 3c**, R = *p*-Cl-C<sub>6</sub>H<sub>4</sub>
- 3d**, R = *p*-C<sub>6</sub>H<sub>5</sub>-C<sub>6</sub>H<sub>4</sub>
- 3e**, R = CH<sub>3</sub>

## EXPERIMENTAL

Melting points were determined in capillary tubes on a Fargo apparatus and are uncorrected. Nuclear magnetic resonance (<sup>1</sup>H and <sup>13</sup>C) spectra were recorded on a Varian XL-GEM 200 spec-

trometer. Chemical shifts were expressed in parts per million ( $\delta$ ) with tetramethylsilane (TMS) as an internal standard. The progress of reaction was followed by thin-layer chromatography (tlc) on silica gel 60 F-254 plates purchased from E. Merck. Flash column chromatography was performed with silica gel 60 (230-400 mesh) from E. Merck. All compounds were analyzed for C, H, and N. The results were within 0.4% of the calculated theoretical values.

#### 7-Phenyl-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3a**).

To a stirred solution of 6-amino-3-oxo-*as*-triazine-5-thione (**1**, 144 mg, 1 mmole) in absolute ethanol (15 ml) at 40° was added 1*N* sodium hydroxide solution (1 ml). While adding sodium hydroxide solution, **1** dissolved, and after several minutes a small quantity of the sodium salt precipitated. The stirred solution was cooled to room temperature then  $\alpha$ -bromoacetophenone (239 mg, 1.2 mmoles) was added slowly under nitrogen atmosphere. The progress of the reaction was monitored by using tlc (methanol:dichloromethane = 1:10). During the reaction period (ca. 20 hours), the pale orange product gradually precipitated. The solid was collected by filtration, washed with ethanol, then recrystallized from methanol to give 167 mg of **2a** as pale yellow fine needles (68%), mp 186-187°; <sup>1</sup>H nmr (DMSO-*d*<sub>6</sub>):  $\delta$  4.90 (s, 1H, vinyl proton), 5.97 (br s, 1H, N8 H), 7.54-8.09 (m, 5H, C<sub>6</sub>H<sub>5</sub>), 11.74 (br s, 1H, N2 H); <sup>13</sup>C nmr (DMSO-*d*<sub>6</sub>):  $\delta$  129.40 (C-6), 129.39, 129.88, 134.81, 136.72 (C<sub>6</sub>H<sub>5</sub>), 143.22 (C-7), 152.24 (C-4a), 166.68 (C-8a), 193.82 (C-3); uv:  $\lambda$  max (nm) 232 (methanol), 225, 232 (0.1*N* sodium hydroxide), 207 (0.1*N* hydrochloric acid).

*Anal.* Calcd. for C<sub>11</sub>H<sub>7</sub>N<sub>4</sub>OS (244.27): C, 54.09; H, 3.30; N, 22.94. Found: C, 54.20; H, 3.69; N, 22.88.

#### 7-(4'-Methoxyphenyl)-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3b**).

This compound was prepared by the same method as **3a** except the pure product was obtained by flash column chromatography (methanol:dichloromethane = 1:5), followed by recrystallization from ethanol to yield 148 mg (54%), mp 194-196°; <sup>1</sup>H nmr (DMSO-*d*<sub>6</sub>):  $\delta$  3.84 (s, 3H, OCH<sub>3</sub>), 4.84 (s, 1H, vinyl proton), 5.93 (br s, 1H, N8 H), 7.07-8.06 (m, 4H, C<sub>6</sub>H<sub>4</sub>), 11.72 (br s, 1H, N2 H); <sup>13</sup>C nmr (DMSO-*d*<sub>6</sub>):  $\delta$  55.92 (OCH<sub>3</sub>), 114.34, 128.72, 131.05, 163.85 (C<sub>6</sub>H<sub>4</sub>), 131.10 (C-6), 142.35 (C-7), 151.33 (C-4a), 165.71 (C-8a), 190.79 (C-3); uv:  $\lambda$  max (nm) 270 (methanol), 232 (0.1*N* sodium hydroxide), 267 (0.1*N* hydrochloric acid).

*Anal.* Calcd. for C<sub>12</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub>S (274.29): C, 52.55; H, 3.67; N, 20.43. Found: C, 52.31; H, 3.79; N, 20.21.

#### 7-(4'-Chlorophenyl)-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3c**).

This compound was obtained as an orange powder, 103 mg (37%), mp 240-241°; <sup>1</sup>H nmr (DMSO-*d*<sub>6</sub>):  $\delta$  4.87 (s, 1H, vinyl proton), 5.95 (br s, 1H, N8 H), 7.63-8.10 (m, 4H, C<sub>6</sub>H<sub>4</sub>), 11.73 (br s, 1H, N2 H); <sup>13</sup>C nmr (DMSO-*d*<sub>6</sub>):  $\delta$  129.22, 130.56, 134.65, 138.90 (C<sub>6</sub>H<sub>4</sub>), 130.60 (C-6), 142.33 (C-7), 151.26 (C-4a), 165.52 (C-8a), 191.74 (C-3); uv:  $\lambda$  max (nm) 232 (methanol), 232 (0.1*N* sodium hydroxide), 259 (0.1*N* hydrochloric acid).

*Anal.* Calcd. for C<sub>11</sub>H<sub>7</sub>N<sub>4</sub>OCl (278.71): C, 47.40; H, 2.53; N, 20.10. Found: C, 47.66; H, 2.37; N, 20.09.

#### 7-(4'-Phenylphenyl)-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3d**).

This compound was obtained by a method similar to that described for **3a** except the reaction was carried out in dry methanol. After recrystallization from methanol and small amount of dimethyl sulfoxide, the pure **3d** separated in 43% yield, mp 215-218°; <sup>1</sup>H nmr (DMSO-*d*<sub>6</sub>):  $\delta$  4.92 (s, 1H, vinyl proton), 5.97 (br s, 1H, N8 H), 7.40-8.16 (m, 9H, C<sub>6</sub>H<sub>4</sub>-C<sub>6</sub>H<sub>5</sub>), 11.76 (br s, 1H, N2 H); <sup>13</sup>C nmr (DMSO-*d*<sub>6</sub>):  $\delta$  127.20, 127.26, 129.24, 129.39, 134.68, 139.06, 145.34 (C<sub>6</sub>H<sub>4</sub>-C<sub>6</sub>H<sub>5</sub>), 128.78 (C-6), 142.38 (C-7), 151.35 (C-4a), 165.63 (C-8a), 192.09 (C-3); uv:  $\lambda$  max (nm) 227 (methanol), 221, 224 (0.1*N* sodium hydroxide), 208 (0.1*N* hydrochloric acid).

*Anal.* Calcd. for C<sub>17</sub>H<sub>12</sub>N<sub>4</sub>OS (320.36): C, 63.74; H, 3.77; N, 17.49. Found: C, 63.85; H, 3.99; N, 17.09.

#### 7-Methyl-8*H*-[1,4]thiazino[2,3-*e*]-1,2,4-triazin-3(2*H*)-one (**3e**).

This compound was obtained as a dark red powder (ethanol), 97 mg (53%), mp 281-282°; <sup>1</sup>H nmr (DMSO-*d*<sub>6</sub>):  $\delta$  1.72 (d, 3, CH<sub>3</sub>, J = 1.2 Hz), 4.65 (q, 1, vinyl proton, J = 1.2 Hz), 9.04 (br s, 1, N8 H), 11.34 (br s, 1, N2 H); <sup>13</sup>C nmr (DMSO-*d*<sub>6</sub>):  $\delta$  19.52 (CH<sub>3</sub>), 81.61 (C-6), 134.53 (C-7), 139.08 (C-4a), 152.80 (C-8a), 168.25 (C-3); uv:  $\lambda$  max (nm) 232 (methanol), 232, 300 (0.1*N* sodium hydroxide), 208, 297 (0.1*N* hydrochloric acid).

*Anal.* Calcd. for C<sub>6</sub>H<sub>6</sub>N<sub>4</sub>OS (182.20): C, 39.55; H, 3.32; N, 30.75. Found: C, 39.48; H, 3.62; N, 31.04.

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#### REFERENCES AND NOTES

- [1] C. C. Cheng, in *Progress in Medicinal Chemistry*, Vol 25, G. P. Ellis and G. B. West, eds, Elsevier Science Publishers, B. V. (Biomedical Division), 1988, pp 35-83.
- [2] C. H. Han, Y. L. Chen, and C. C. Tzeng, *Nucleosides Nucleotides*, **10**, 1390 (1991).
- [3] K. H. Lee, Y. L. Chen, B. R. Huang, Q. Y. Zhu, T. C. Chou, and C. C. Tzeng, *Nucleosides Nucleotides*, **10**, 1407 (1991).
- [4] C. C. Tzeng, U. Rychlewsha, D. J. Hodgson, and R. P. Panzica, *J. Heterocyclic Chem.*, **23**, 33 (1986).
- [5] J. Riand, C. C. Tzeng, M. T. Chenon, and R. P. Panzica, *J. Chem. Soc., Perkin Trans. 2*, 931 (1986).
- [6] C. J. Cheer, S. Kokkou, C. C. Tzeng, and R. P. Panzica, *J. Chem. Soc., Perkin Trans. 2*, 1455 (1987).
- [7] D. Falke and B. Rada, *Acta Virol.*, **14**, 115 (1970).
- [8] R. W. Sidwell, G. J. Dixon, S. M. Sellers, and F. M. Schabel Jr., *Appl. Microbiol.*, **16**, 370 (1968).
- [9] W. A. Creasey, M. E. Fink, R. E. Handschumacker, and P. Calabresi, *Cancer Res.*, **23**, 444 (1963).
- [10] T. R. Walters, R. J. A. Aur, K. Hernandez, T. Vietti, and D. Pinkel, *Cancer*, **29**, 1057 (1963).
- [11] G. Matolcsy, *Acta Phytopathol.*, **1**, 245 (1966).
- [12] C. C. Tzeng, N. C. Motola, and R. P. Panzica, *J. Org. Chem.*, **48**, 1271 (1983).
- [13] Y. H. Kim and H. G. Mautner, *J. Med. Chem.*, **17**, 369 (1974).
- [14] M. G. Nair, L. H. Boyce, and M. A. Berry, *J. Org. Chem.*, **46**, 3354 (1981) and references therein.
- [15] R. N. Henrie II, R. A. Lazarus, and S. J. Benkovic, *J. Med. Chem.*, **26**, 559 (1983).